

# Recovery of spheroidized graphite from spent lithium ion batteries

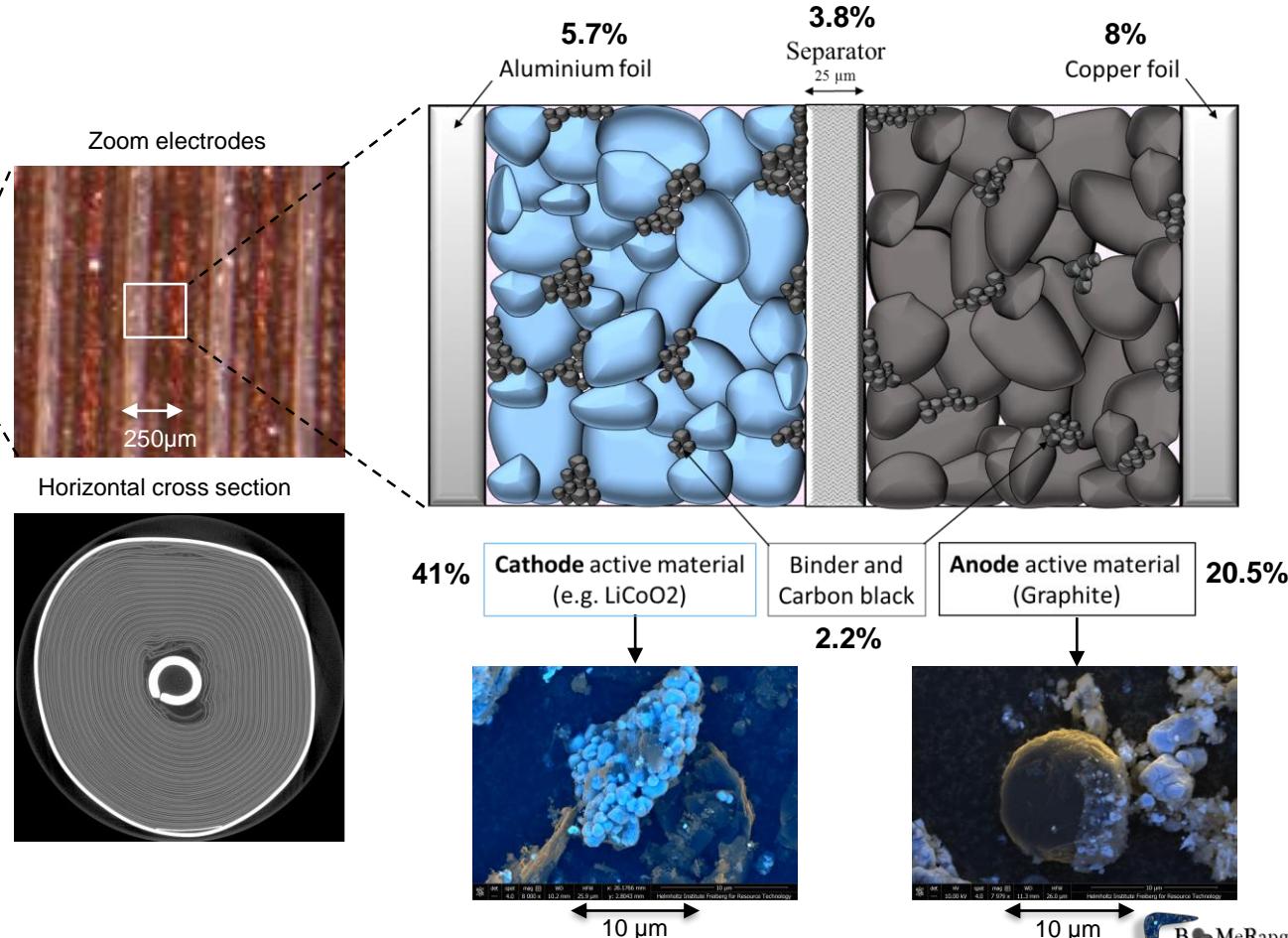
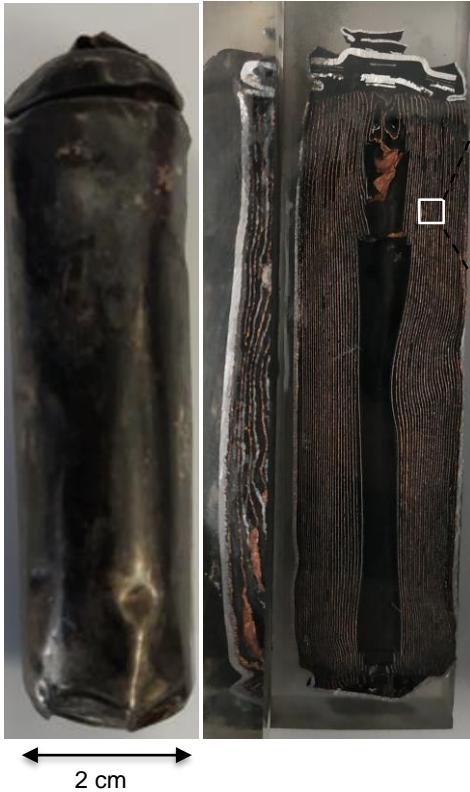
Anna Vanderbruggen, Martin Rudolph

# Outline



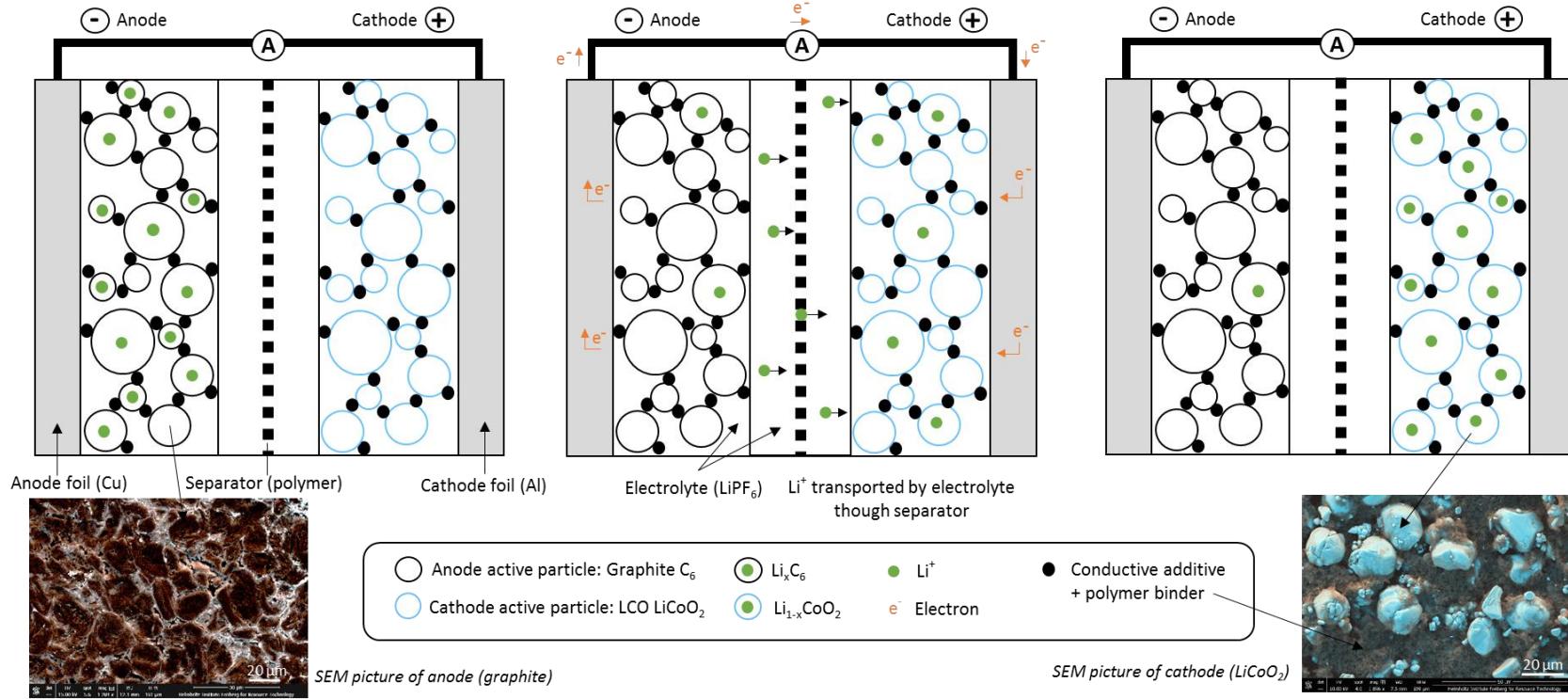
# Motivation

Pyrolysed battery Vertical cross section



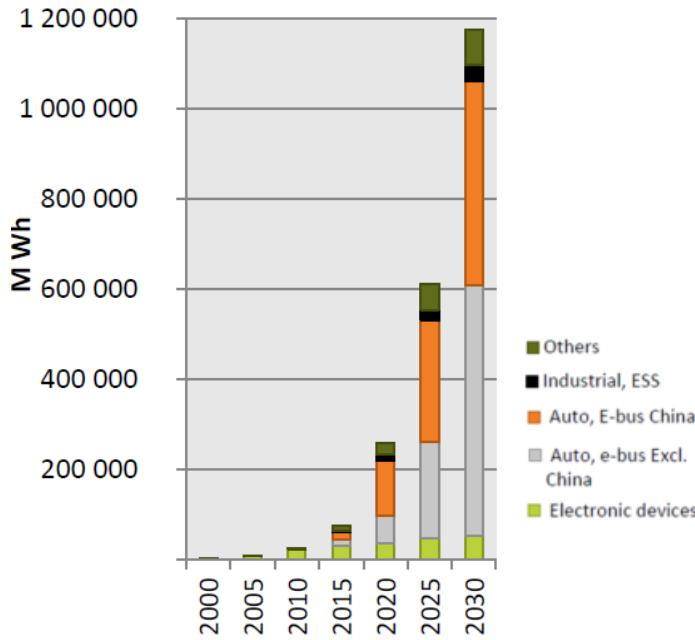


Reactions during the discharge:

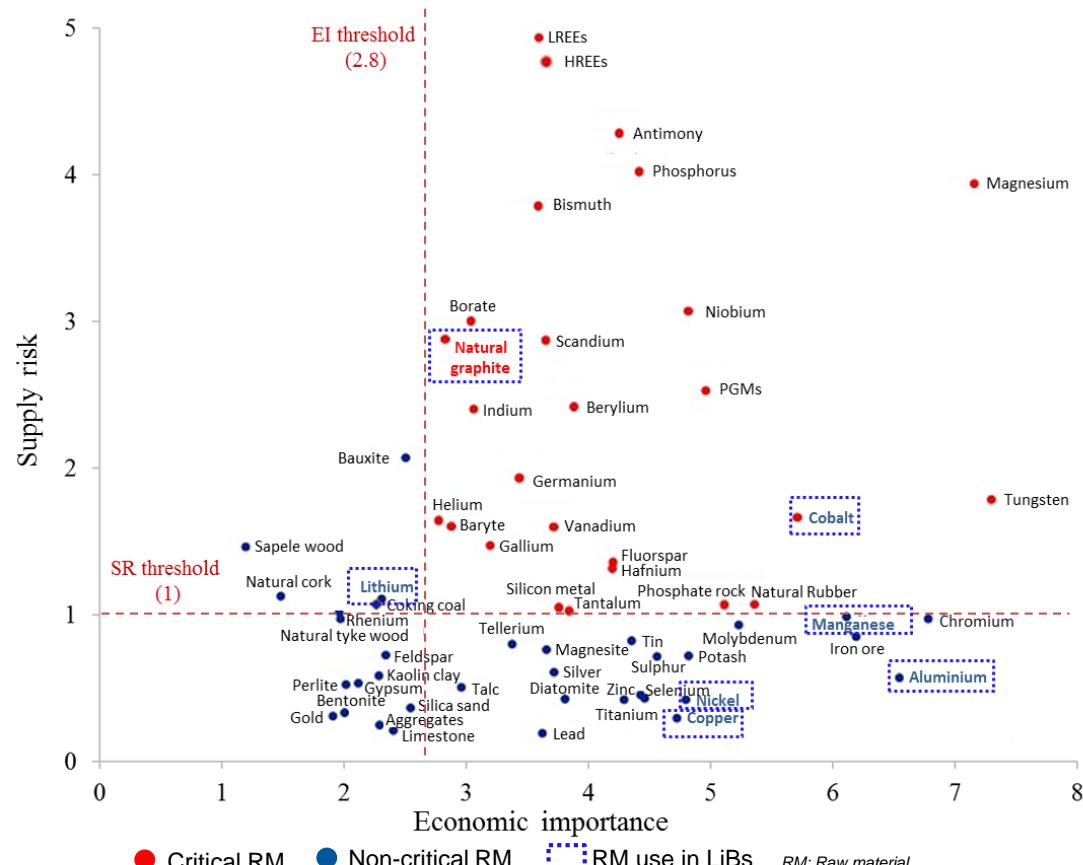


# Motivation

## Why recycle LiBs?



Li-ion battery sales, MWh, Worldwide, 2000-2030  
(C. Pillot at ICBR 2019, AVICENNE Energy)



Critical Raw Material CRM from European Commission  
(EC report, 2018)

# Motivation

## Actual recycling processes

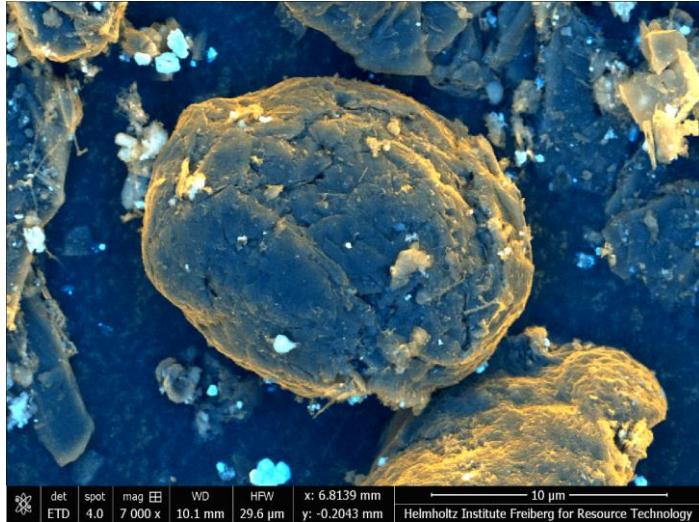
Company/Process	Capacity (t/y)	Technology	Main products	Graphite
<b>Accurec GmbH</b>	6 000	Pyro-dominant	Co alloy, $\text{Li}_2\text{CO}_3$	Not recovered
<b>Akkuser Ltd</b>	4 000	Pyro-dominant	Metal powder	Remain as fraction of the black mass
<b>Sumitomo-Sony</b>	150	Pyro-dominant	Co alloy, Co metal	Lost during calcination
<b>Umicore</b>	7 000	Pyro-dominant	Ni-Co alloy, $\text{NiCo}_3$ , $\text{NiSO}_4$ , $\text{CoCO}_3$ , $\text{CoSO}_4$	Lost, used as a reduction agent
<b>Duesenfeld</b>	4000	Physical and Hydro	Organic carbonates, $\text{LiPF}_6$ , Fe metal, Cu metal, Al metal, graphite, cathode material	Aiming to recover for LiB purposes
<b>Recupyl</b>	110	Physical and Hydro	$\text{Li}_2\text{CO}_3$ , Co metal	Filtered off in leaching step
<b>OnTo</b>	Nd	Physical and Hydro-dominant	NMC cathode active material	Aiming to recover for LiB purposes
<b>TOXCO (Retrieval)</b>	4 500	Hydro-dominant	$\text{Li}_2\text{CO}_3$ , mixed metal oxides	Recovered during filtration as MeO-graphite cake

Summary of the actual major LiB recycling process, modified from (Liu et al., 2019) and (Velázquez-Martínez et al., 2019)

# Motivation

## Why recycle graphite?

- 15-20 wt. % of LiB is graphite  
Usually ending in the process waste!
- Graphite critical raw material in:
  - Europe
  - USA
  - Australia
- Flake graphite production:  
China 69% **Europe 3%**
- Increase recycling rate of LiB  
Directive 2006/66/EC: 50 wt.%



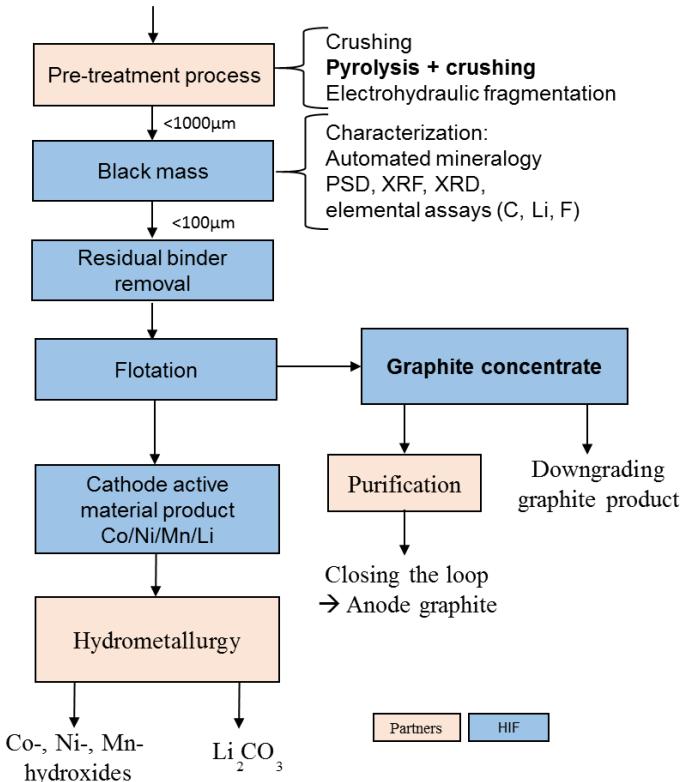
### Spheroidized graphite

- 9000-10000 \$/t
- China: 90 % production

(Hatch, 2016;  
Frey, 2018;  
CRM reports, 2019)

# BooMeRang project

Spent lithium ion batteries



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RESEARCH FOR GRAND CHALLENGES



## Research objectives:

- Study the influence of different recycling pre-treatment on the electrode active material liberation
- Recovering spheroidized graphite by froth flotation

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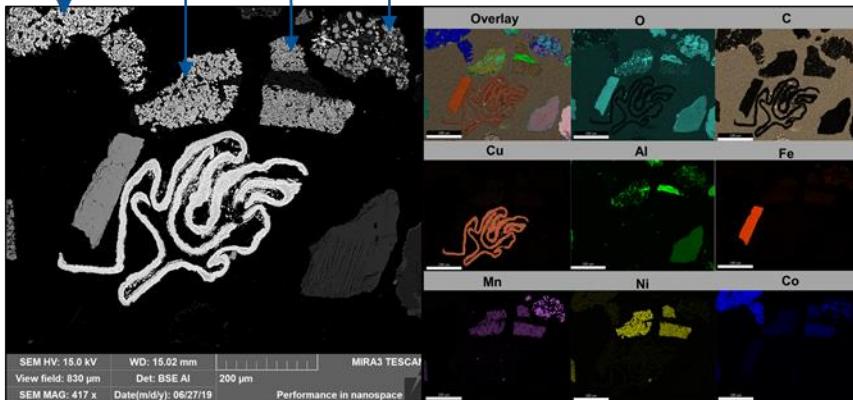
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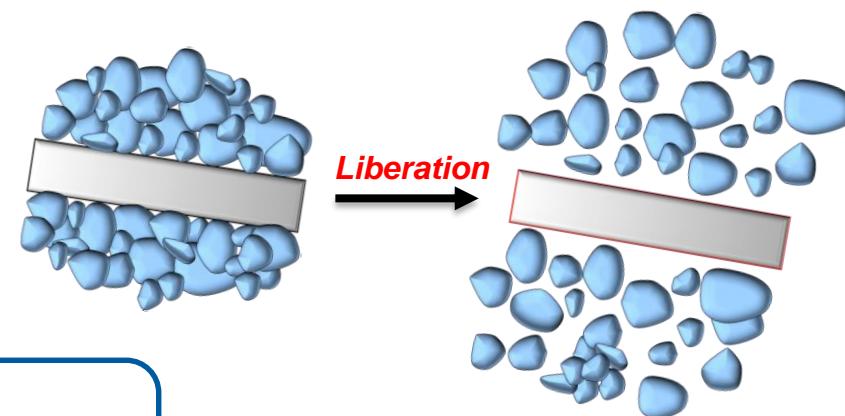
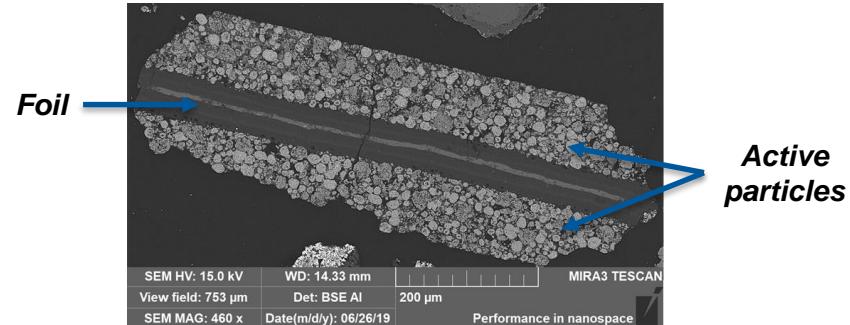
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# Characterization: Black mass

LCO      NCA      LMO      NMC  
LiCoO<sub>2</sub>      LiNiCoAlO<sub>2</sub>      LiMn<sub>2</sub>O<sub>4</sub>      LiNiCoMnO<sub>2</sub>

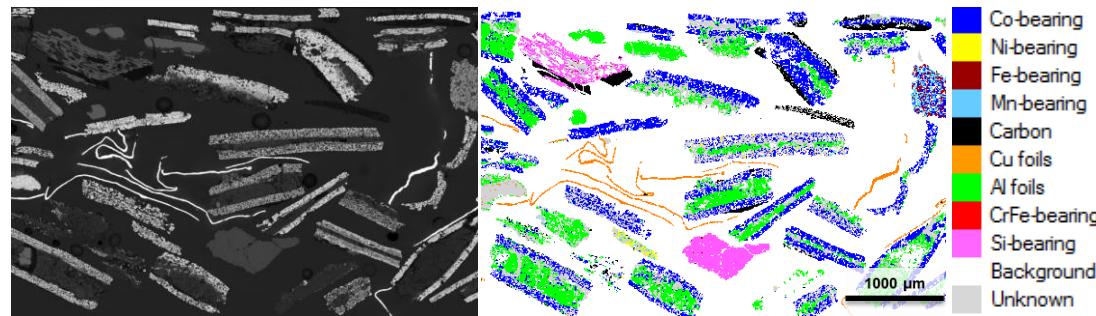


*Elemental mapping of coarse fraction of the black mass with hypothesis for cathode active particle chemistry*



- Process able to deal with different battery chemistry
- Recycling efficiency depends on **particle liberation**

# Characterization: *Black mass coarse fraction*

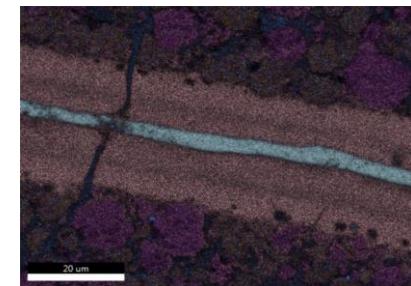
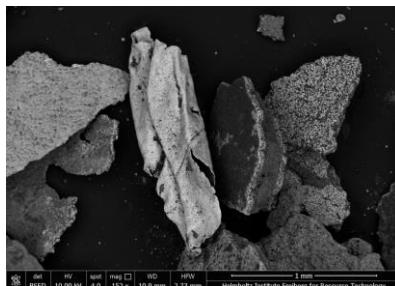
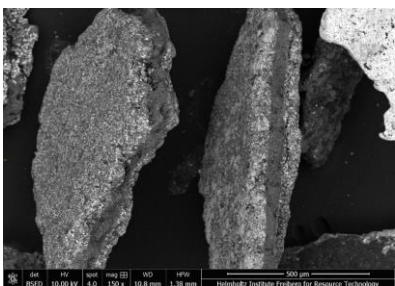
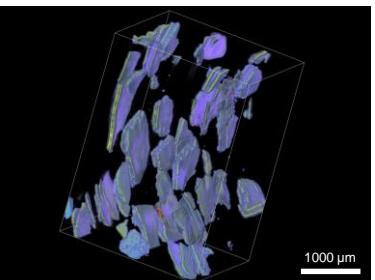


Automated mineralogy:  
SEM (BSE images) + EDX analysis  
**Challenging for secondary material!**

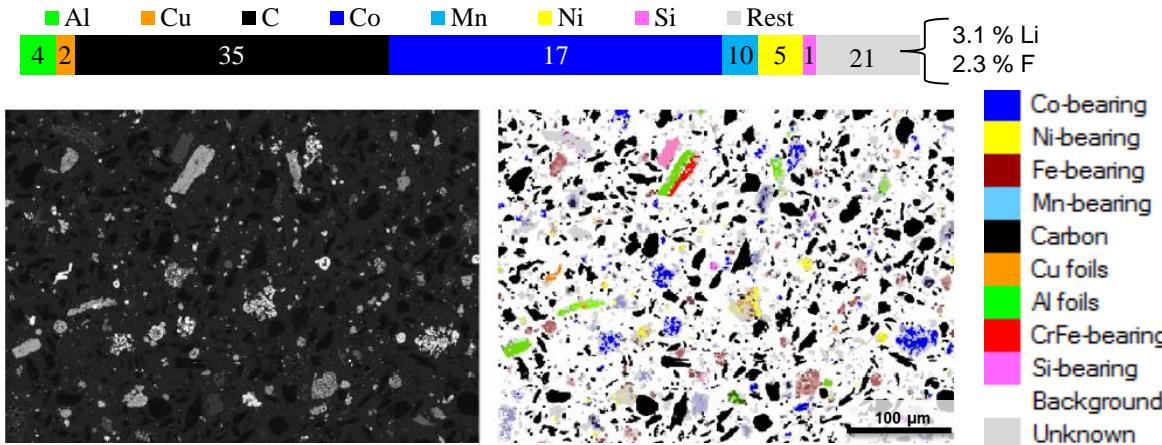
Estimation of the liberation degree:

- 35 wt. % of the aluminium foil
- 92 wt. % of the copper foils are liberated

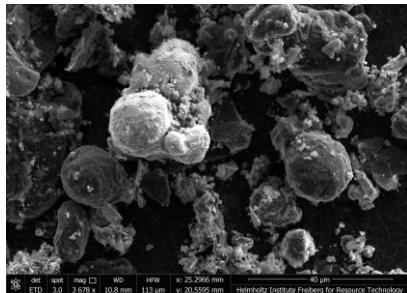
Able to understand and quantify the losses



# Characterization: *Black mass fine fraction*



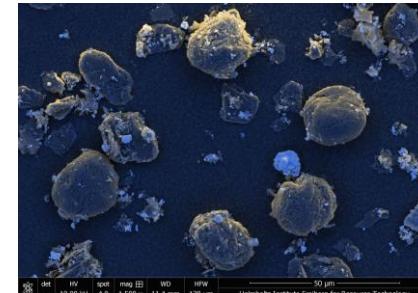
BSE image and processed MLA image of the  $< 63 \mu\text{m}$  fraction



Fraction below 100 µm



Liberated  $\text{LiCoO}_2$  particle



Liberated graphite particles

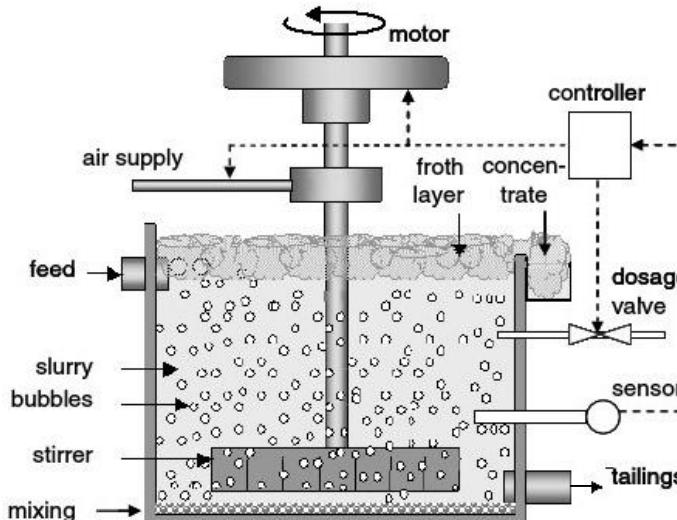
- Graphite liberated: 10-30 µm
- Spheroidized shape is conserved

# Flotation theory

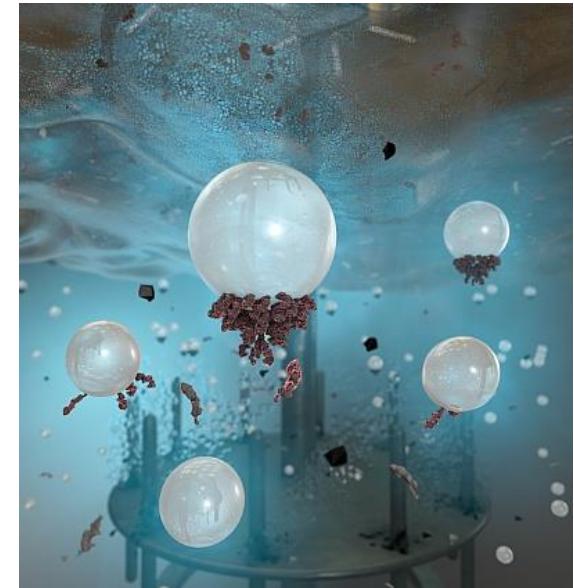
**Flotation:** separation technique based on physico-chemistry surface properties

→ Liberated particles + surface wettability difference

A 2L Outotec cell (GTK labCell)

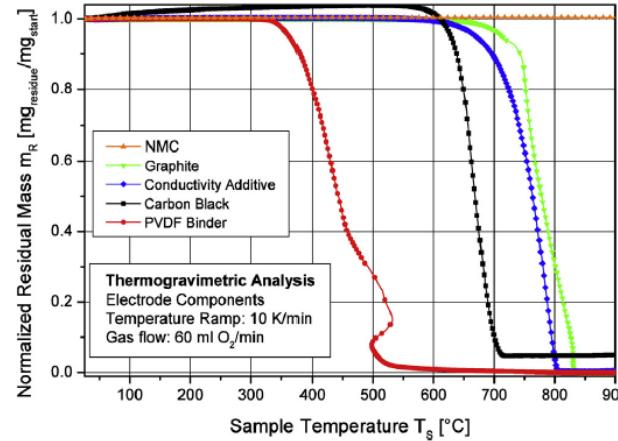
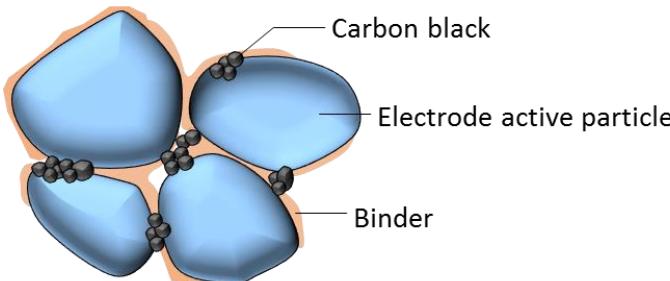


Froth flotation diagram  
(Kramer et al., 2012)



3D visualization of flotation

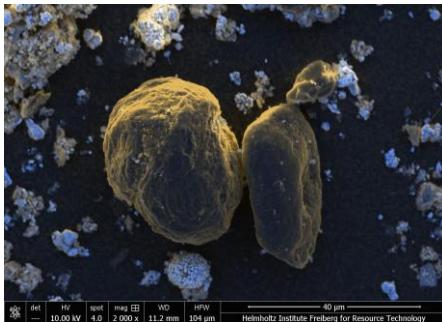
# Surface properties



Decomposition of the binder during pyrolysis

→ Binder residuals

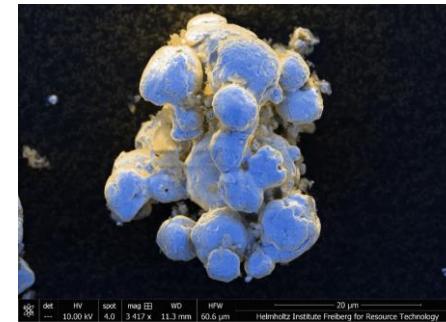
Thermogravimetric analysis of single electrode components (Hanisch et al., 2015)



Graphite particles  
Hydrophobic



Binder residual  
Hydrophobic

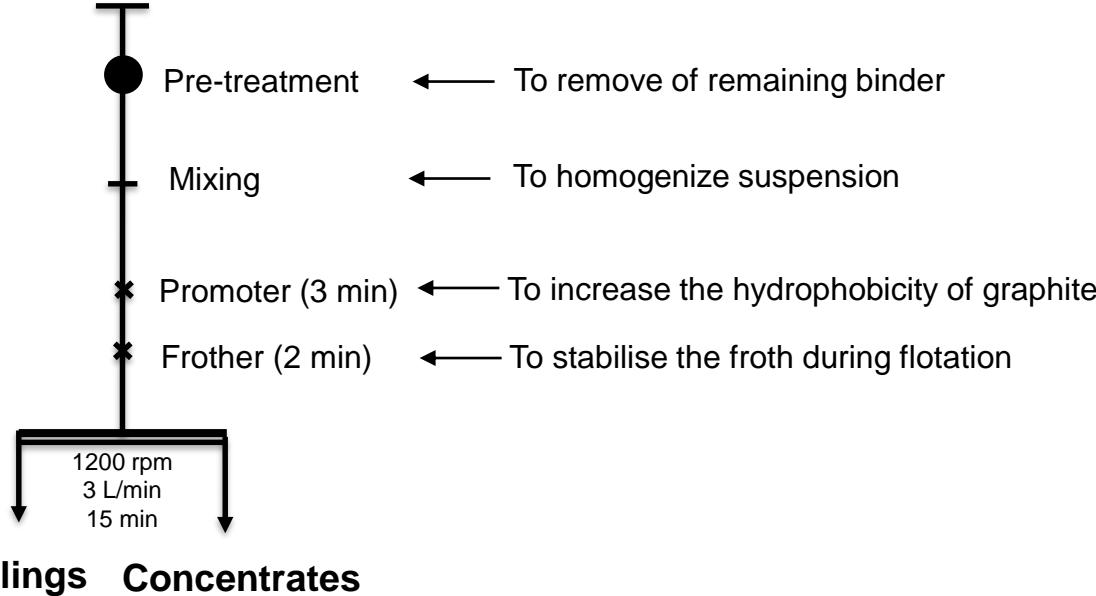


Cathode active particle  
Hydrophilic

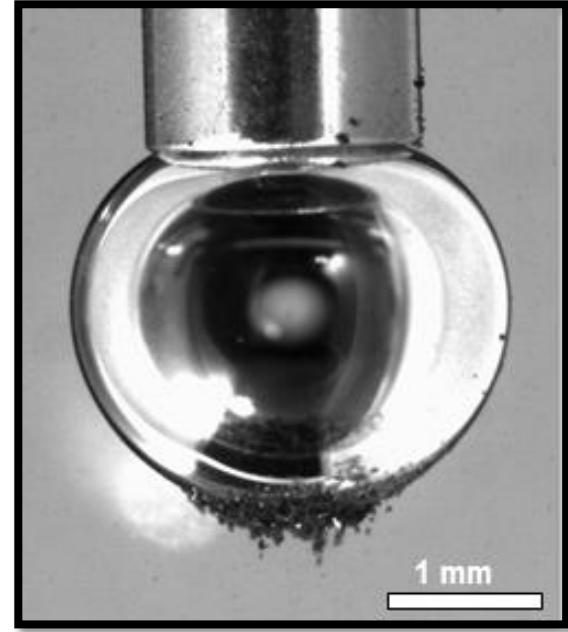
Difference of wettability → Flotation

# Flotation methods

## Black mass Flotation feed

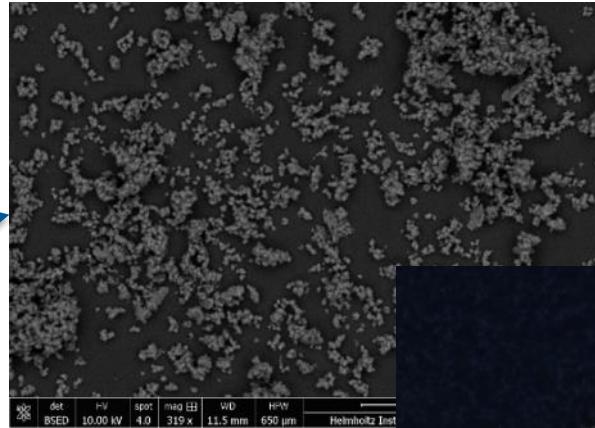


Simplified black mass  
flotation flowsheet



Black mass in contact with  
air bubble in water

# Flotation: Results and Discussion

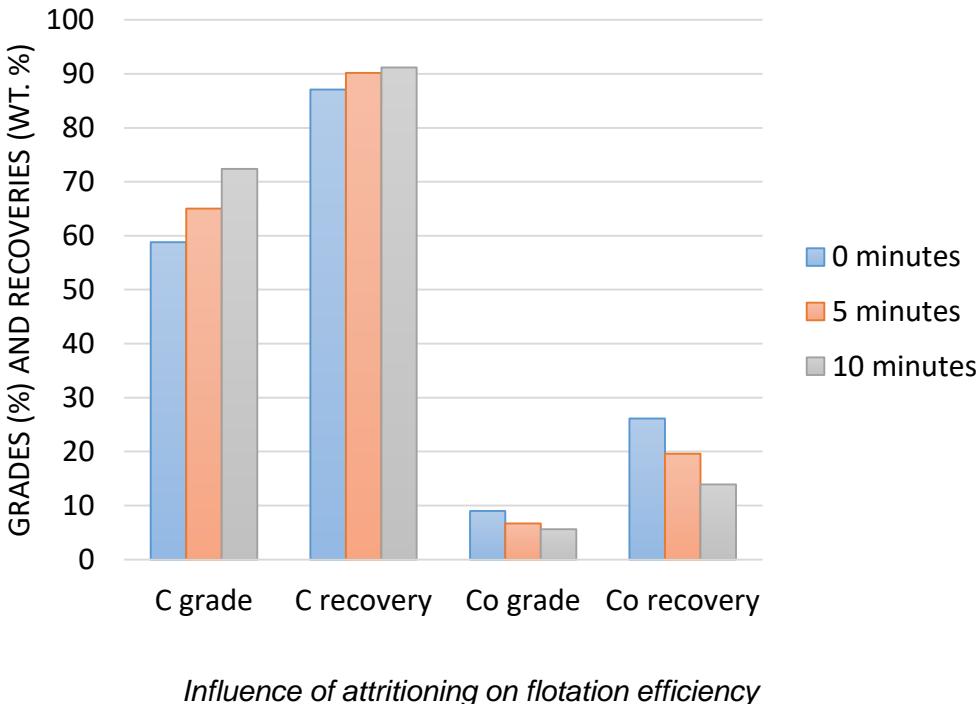


After attritioning, floating  
deaggregated particles and  
black mass in sediment

→ Hydrophobic residual binder



# Flotation: Results and Discussion



## Attritioning pre-treatment:

- Remove a part of binder:
  - Should decrease true flotation
- Deaggregate the cathode active particles
- Disperse the fine particles ( $d_{50}$  12  $\mu\text{m}$ )
- Decrease entrainment mechanism

*Initial froth forming*



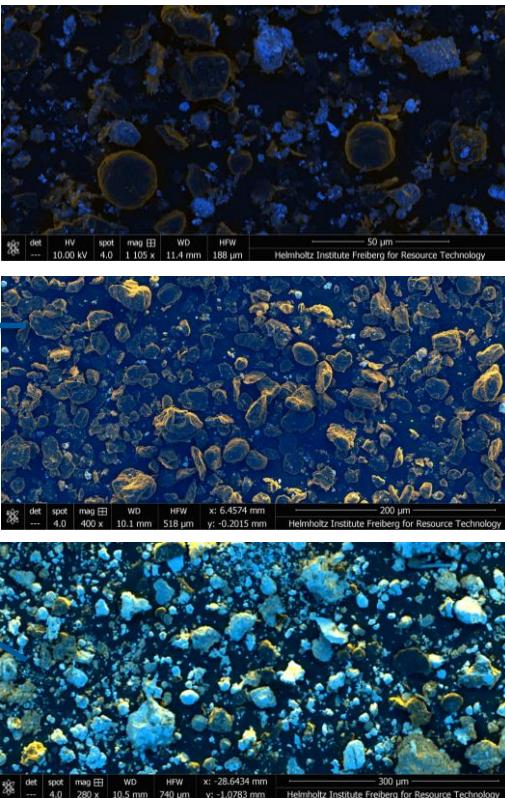
20 min attritioning  
better drainage  
→ Less entrainment

No attritioning

# Flotation: Results and Discussion



Rougher flotation



SEM images in false color of flotation products:  
Feed (top), concentrate (middle), tailings (bottom)

## Flotation feed:

35 wt.% C  
17 wt.% Co  
10 wt.% Mn  
3.1 wt.% Li

## Graphite concentrate:

74-79 wt.% C with 94-96% recovery

## Metals product:

28 wt.% Co with 80% recovery  
11 wt.% Mn with 70% recovery  
8 wt.% Ni with 65% recovery  
2.4 wt.% Li with 77% recovery

# Summary and Outlook

- ✓ It is possible to **recover spheroidized graphite** from Black Mass using flotation after pyrolysis and attritioning (98% Recovery, 79% Grade)
- ✓ With a **leaching post-treatment** it will be possible to obtain anode grade graphite (~96% Recovery, >92% Grade)
- ✓ The flotation “tailings” are **further hydrometallurgically treated** for the resource efficient recovery of **Li, Co, Ni, Mn...**
- We will focus now on the **battery pre-treatments steps** (mechanical and thermal) and its influence on the **surface properties** of the very fine black mass anode and cathode particles

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# Thank you for your attention!



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A!  
Aalto University

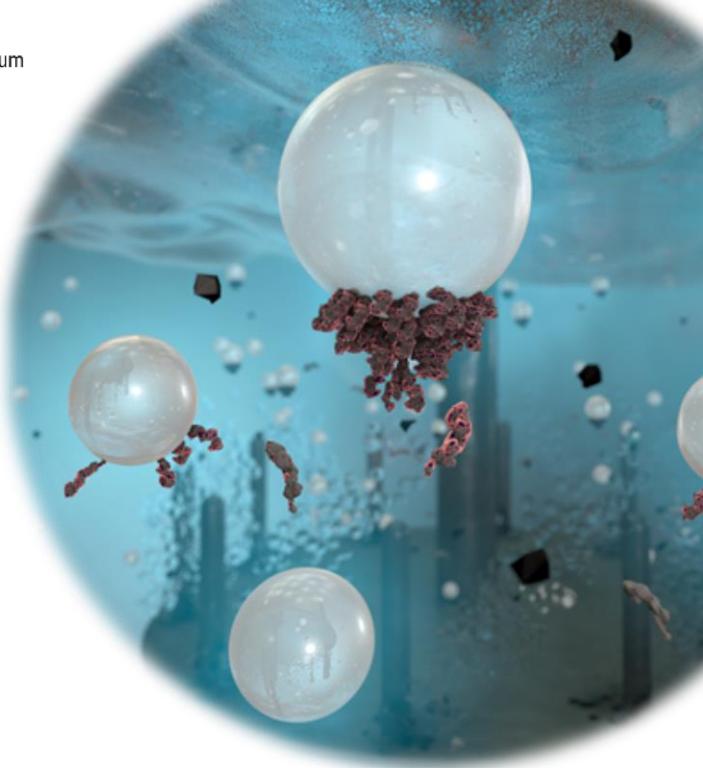
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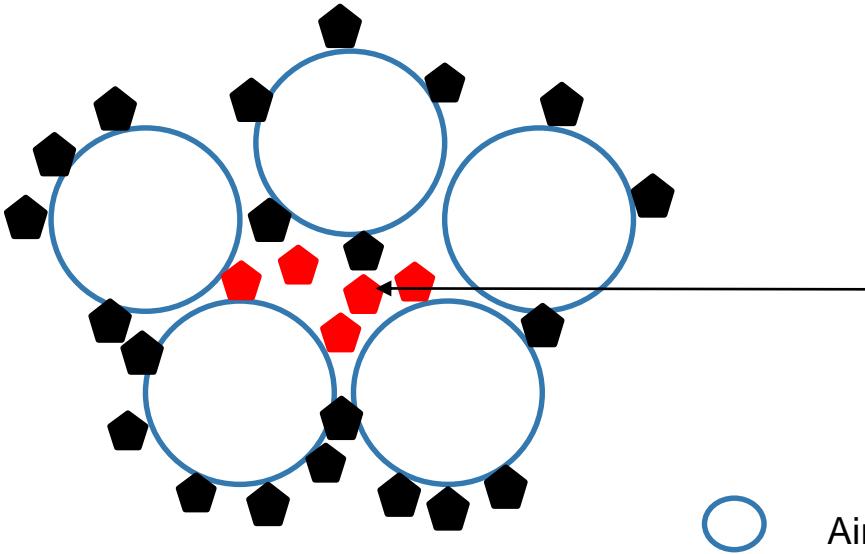
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Critical raw material reports:

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<https://www.industry.gov.au/sites/default/files/2019-03/australias-critical-minerals-strategy-2019.pdf>
- Critical Minerals and U.S. Public Policy, June 2019.  
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- Critical Raw Materials - European Commission. January 2018  
[https://ec.europa.eu/growth/sectors/raw-materials/specific-interest/critical\\_en](https://ec.europa.eu/growth/sectors/raw-materials/specific-interest/critical_en)

# Example of entrainment in flotation



Entrapment of hydrophilic particles  
→ Particles carried up  
into the concentrate



Air bubble



Hydrophobic particle (e.g Graphite)



Hydrophilic particle (e.g Metal)